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Preparation and Crystal Structure of the Dinuclear, Asymmetric Dioxo Complex $(\eta^{5}-C_{5}Me_{5})(CO)_{3}W-W(O)_{2}(\eta^{5}-C_{5}Me_{5})$

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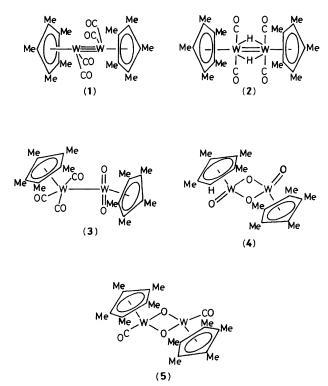
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The dinuclear, asymmetric, dioxo complex $(\eta^5-C_5Me_5)(CO)_3W-W(O)_2(\eta^5-C_5Me_5)$ (3) can be prepared by air oxidation of $[(C_5Me_5)W(CO)_2]_2$ (1) or $(C_5Me_5)W(CO)_2(\mu-H)]_2$ (2); in this compound W^I is directly connected with W^V

Presently the various chemistry of oxo complexes is attracting great attention and application.^{1,2} An easy approach to oxo complexes is the exposure of carbonyl complexes to air. In this manner the dinuclear complexes $[(C_5Me_5)W(CO)_2]_2$ (1) or $[(C_5Me_5)W(CO)_2(\mu-H)]_2$ (2) react with air to yield the title compound (3).

This result is not that of the reported reaction of (1) and oxygen to give complexes (4) and (5) as products.¹ We could not detect these species in our experiments. The Cr analogue $[(C_5Me_5)Cr(O)(\mu-O)]_2$, however, has been described.³ Mononuclear dioxo complexes of tungsten and molybdenum con-taining cyclopentadienyl (Cp) ligands are known, *e.g.*



 $CpW(O)_2CH_2SiMe_{3,4}$ $CpW(O)_2Fc$ (Fc = ferrocenyl),⁵ and $CpMo(O)_2Cl.^6$ We are not aware of any other dinuclear dioxo complex in which W¹ is directly connected with W^V.

The formation of (3)[†] involves complexation of oxygen and elimination and uptake of carbon monoxide. The $(C_5Me_5)W(CO)_3$ fragment is isolobal with Me There are two independent molecules of (3) in the asymmetric unit. The two molecules are nearly identical with the exception of a rotational difference in the orientation of the η -C₅Me₅ ligand bonded to the W¹(CO)₃ fragment. The molecular geometry of one of these molecules is depicted in Figure 1. The $(C_5Me_5)W(CO)_3$ fragment is connected to the $(C_5Me_5)WO_2$ unit via a W–W bond. The two metal-metal distances in (3) [av. 2.877(3) Å] are shorter than the one in the dinuclear carbonyl complex [CpW(CO)₃]₂⁷ or in the fulvalene complex $C_{10}H_8W_2(CO)_6$ [3.347(1) Å],⁸ but they are still in the range of literature values for W–W single bonds [see ref. 8(26)]. Bond

† Spectroscopic data: ¹H n.m.r. (90 MHz, [²H₆]acetone, -20 °C) δ 2.02 (s) and 2.08 (s); ¹³C n.m.r. (22.5 MHz, -20 °C) δ 103.5 (s), 119.5 (s, C_5Me_5), 10.9 (s), 12.2 [s, C_5Me_5 , ¹J(C,H) 127.9 Hz]; 224.2 and 223.6 (CO); i.r.: $v_{(CO)}$ 1962(s), 1884(vs) (tetrahydrofuran); $v_{(W=O)}$ 936, 894 cm⁻¹ (KBr); *m/z* 726 (*M*⁺, rel. ¹⁸⁴W).

Acceptable elemental analyses were obtained for C and H; m.p. (decomp.) 122 $^{\circ}$ C.

Crystal data: $C_{23}H_{30}O_5W_2$, M = 754.2, space group $P2_1/c$, a = 29.106(5), b = 10.418(4), c = 15.703(4) Å, $\beta = 105.66(2)^\circ$, Z = 8, $D_c = 2.19$ g cm⁻¹. Intensity data were obtained at -150 °C by 2θ - θ scans on an Enraf-Nonius CAD4 diffractometer with graphite monochromated Mo- K_{α} radiation ($\lambda 0.71073$ Å). The structure was solved by heavy atoms techniques. Least-squares refinement of 6088 independent observed reflections $[F_o \ge 5\sigma(F_o)]$ gave R = 0.062. Due to crystallographic difficulties resulting from an irregularly shaped crystal and a high value of μ (96.3 cm⁻¹), only the tungsten atoms were refined anisotropically. Scattering factors for neutral atoms and anomalous dispersion corrections were taken from ref. 9. Data were corrected for absorption. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

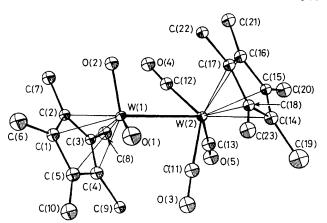


Figure 1. ORTEP¹⁰ drawing of one of the two unique molecules of (3). Ellipsoids represent 50% probability surfaces. Selected average distances (Å) and angles (degrees) with estimated standard deviations [Z(1) and Z(2) are the centroids of the rings containing C(5) and C(15), respectively]. WV–Z(1) 2.14, WU–CO 1.95(2), WU–Z(2) 2.00 Å; WU–WV–O 100(1), O–WV–O 106(2), WV–WU–CO [corresponding to C(11) and C(12)] 70(1), WV–WU–CO [corresponding to C(13)] 120.1(4), OC–WU–CO [corresponding to C(11)–W(2)–C(13)] 84(2), OC–WU–CO [corresponding to C(11)–W(2)–C(12)] 124(4)°.

distances and bond angles of the $(C_5Me_5)W(CO)_3$ moiety are comparable to those in the CpW(CO)₃ unit in $[CpW(CO)_3]_2$,⁷ except that the two carbonyl ligands of each molecule in (**3**) are bent towards the other tungsten atom occupying a semibridging position. The four such W–C contacts average 2.87(3) Å. The W^v=O distances [av. 1.73(2) Å] are in the same range as the corresponding metal–oxygen bond lengths [1.716(5) and 1.723(5) Å] in the monomolecular complex CpW(O)₂CH₂SiMe₃.⁴

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